

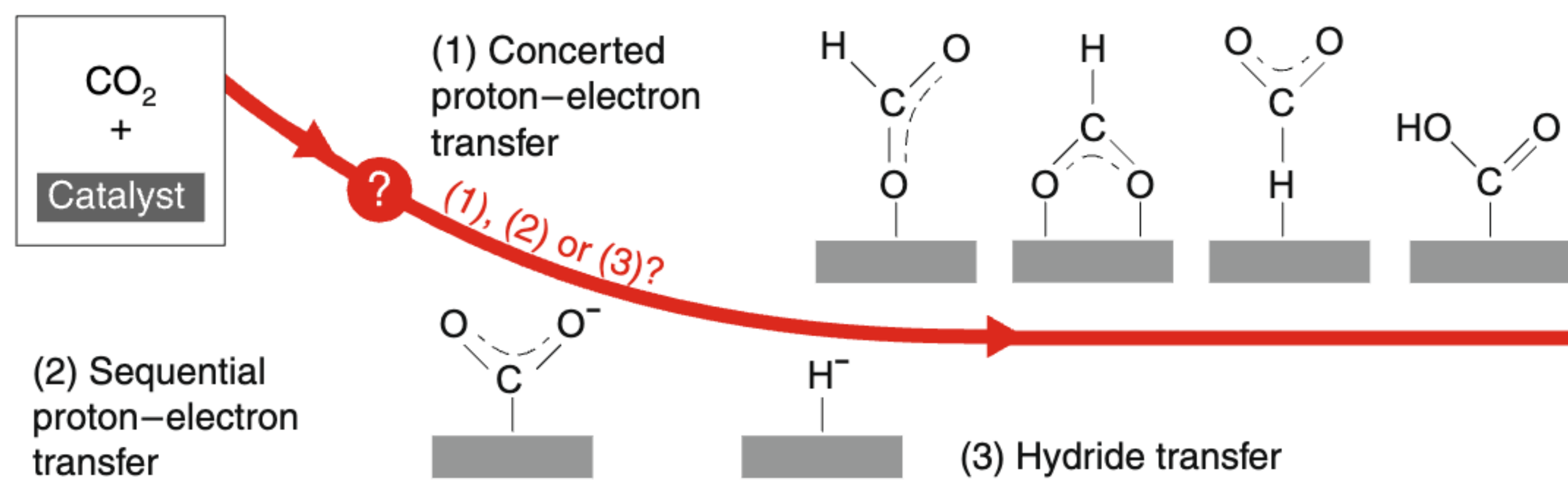
Inverted Region in Electrochemical Reduction of CO₂ Induced by Potential-dependent Pauli Repulsion

Leyu Liu, Hai Xiao*

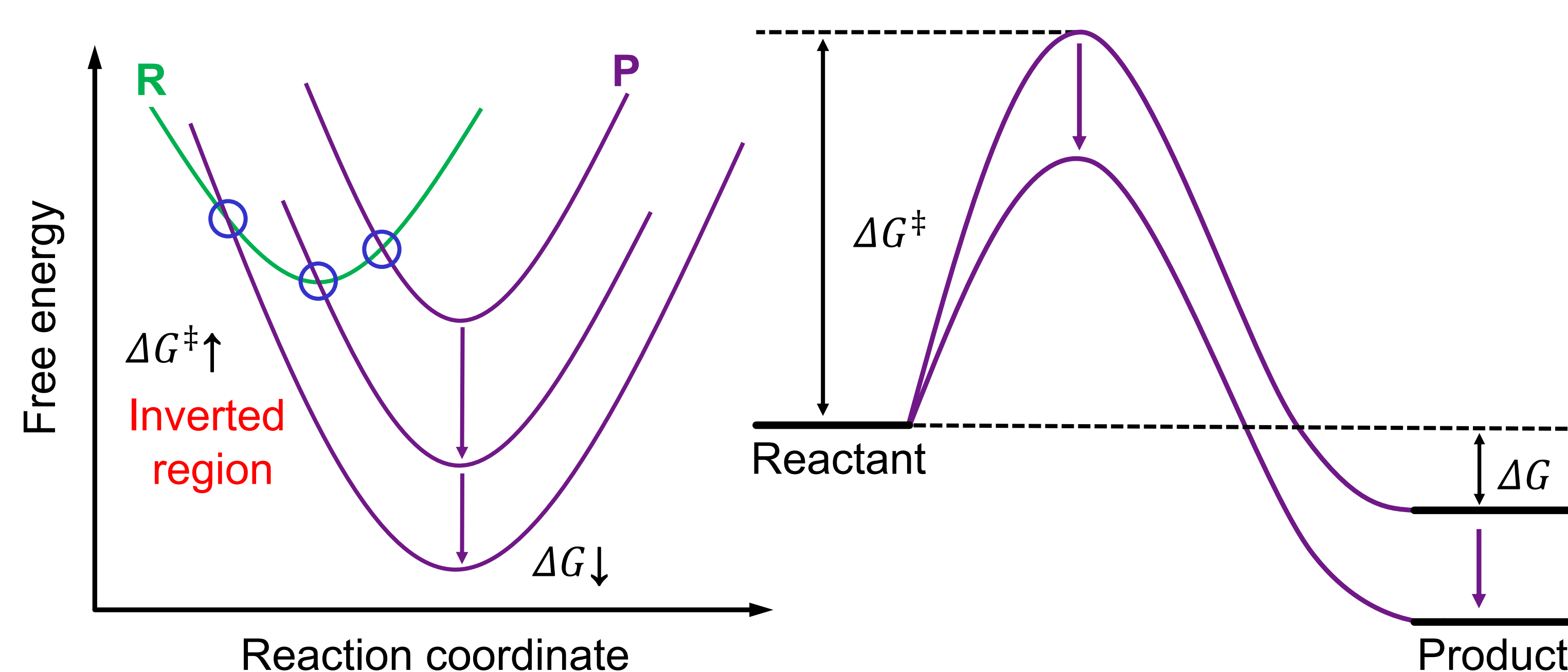
Department of Chemistry, Tsinghua University, Beijing 100084 Email:liu-ly19@mails.tsinghua.edu.cn

Introduction

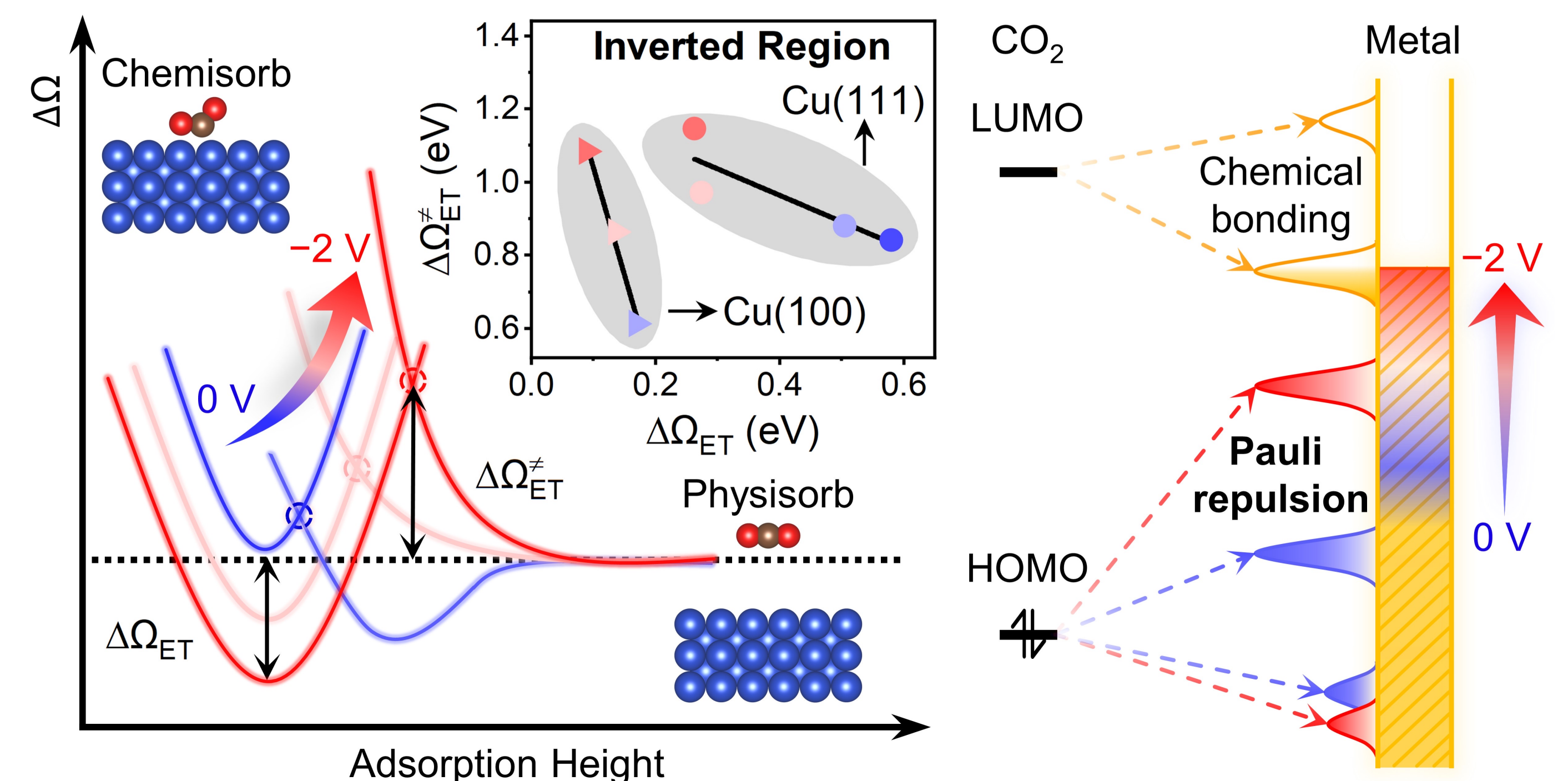
CO₂ activation: SEPT vs. CPET



Electron transfer kinetics: Marcus theory vs. BEP relation



Abstract

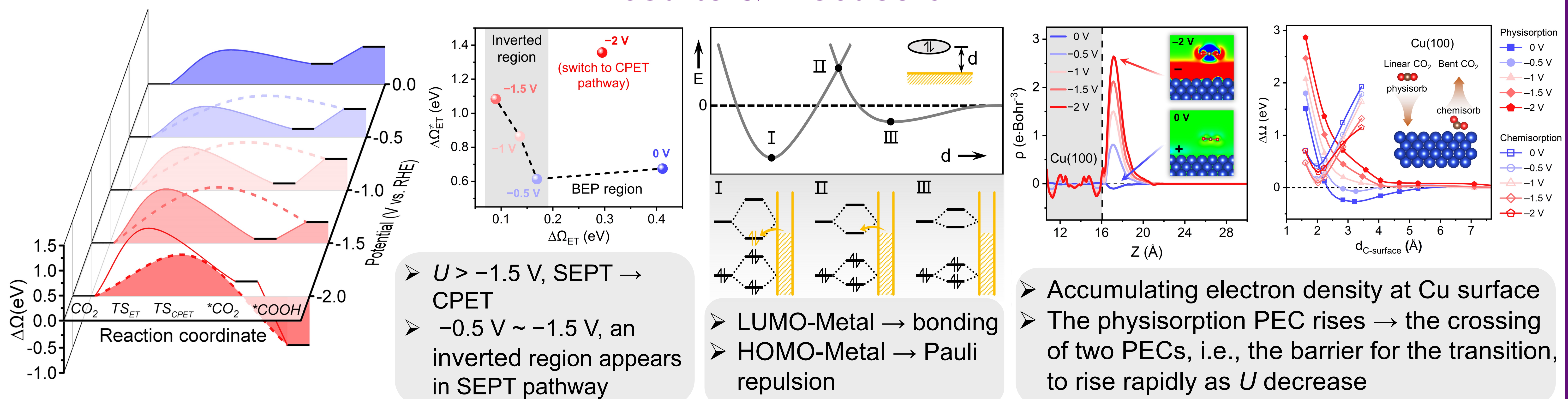


- The CO₂ activation mechanism in eCO₂RR varies with U
- The barrier of the electron-transfer step in the SEPT mechanism exhibits an inverted region as U decrease

Methods

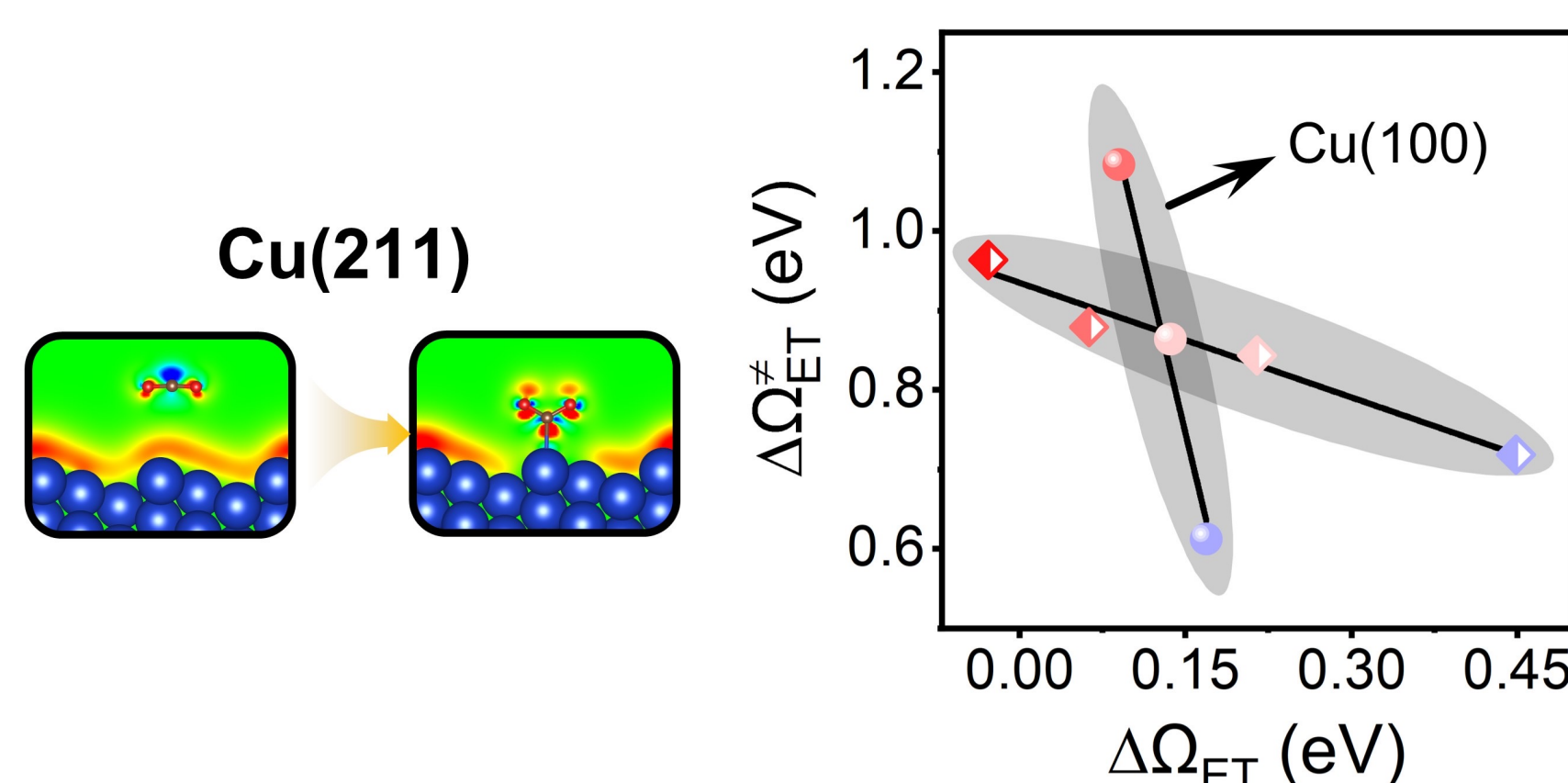
- Grand canonical density functional theory (GC-DFT)
- ASE - Atomic Simulation Environment
- JDFTx - software for joint density-functional theory

Results & Discussion

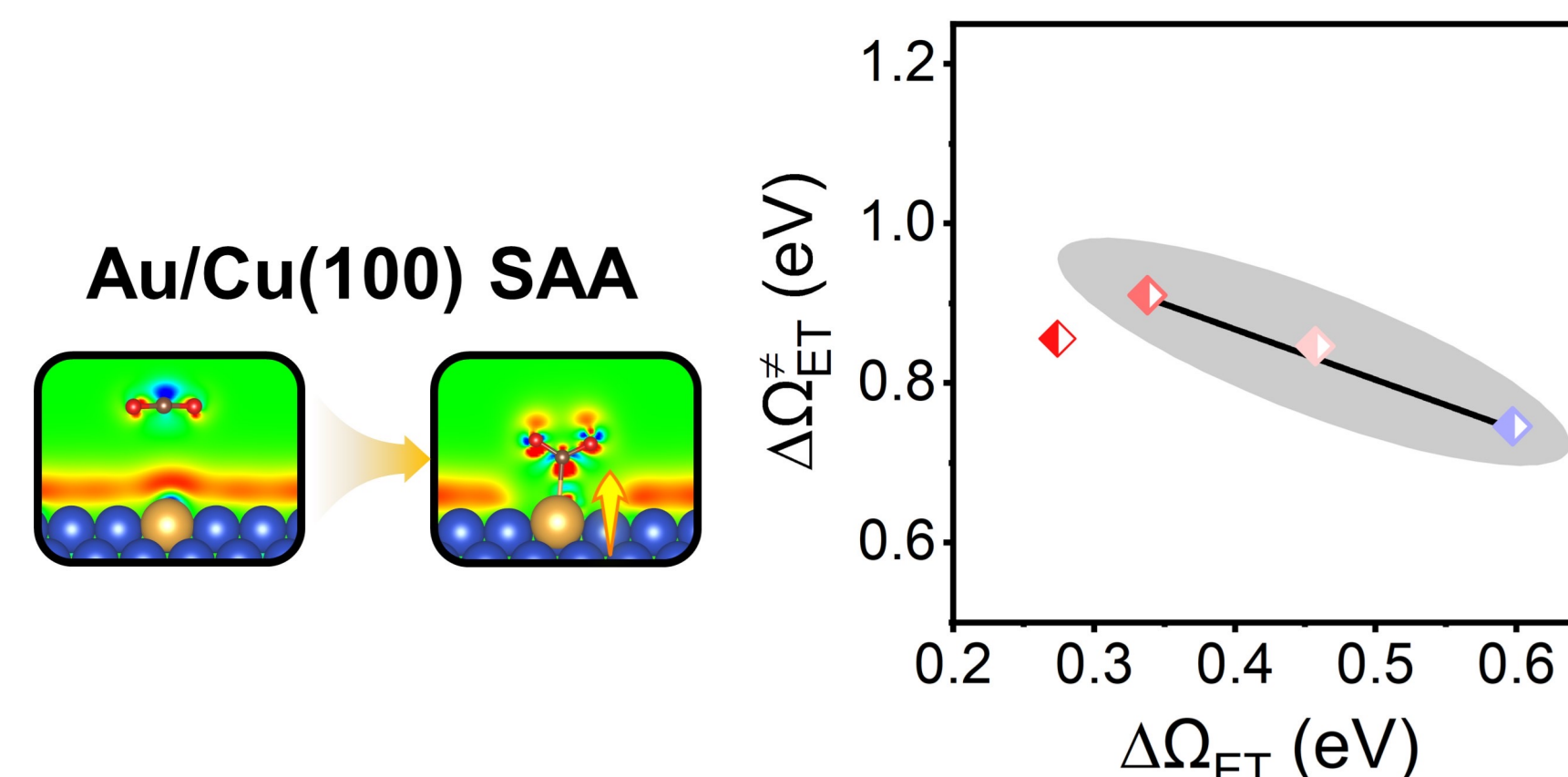


➤ Catalyst design

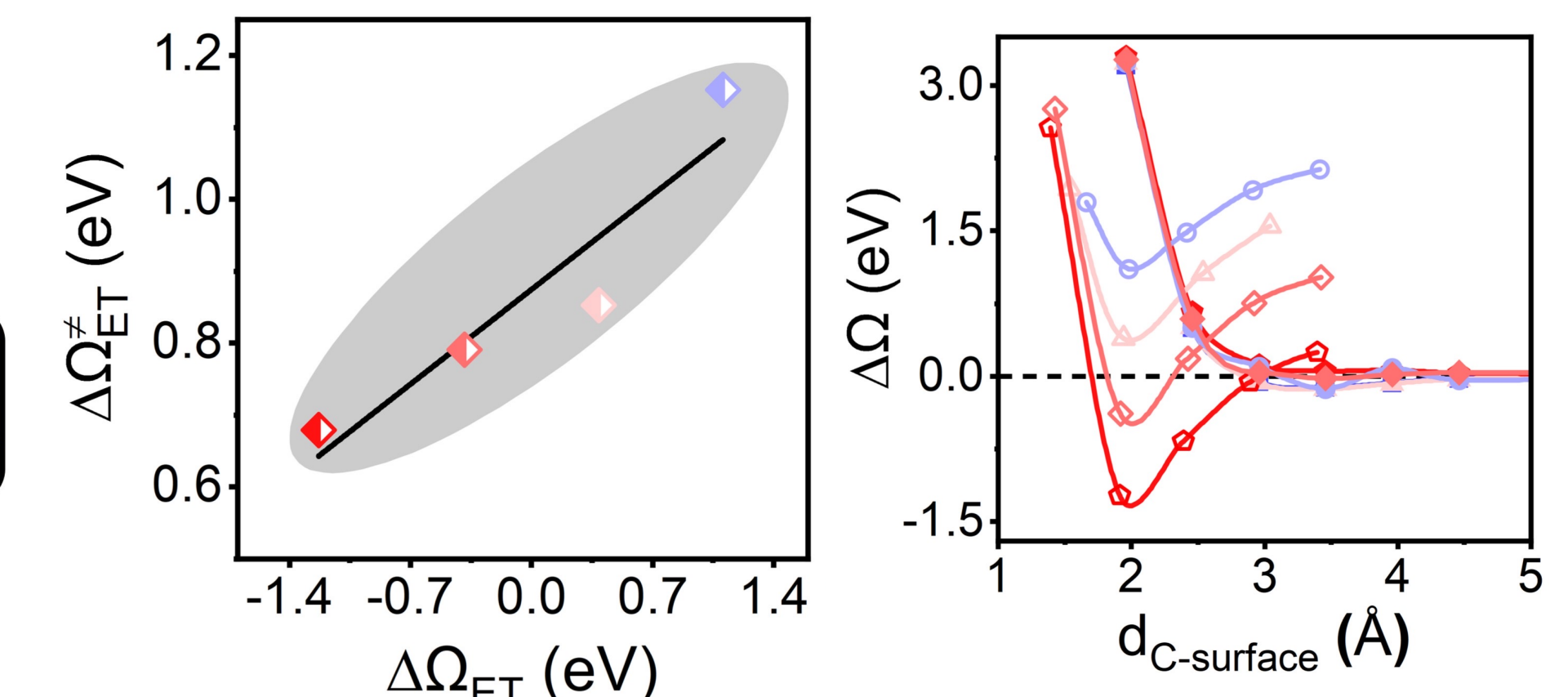
1. Increasing surface roughness



2. Increasing the coordination flexibility of the active sites



3. a substrate with less charging



Conclusion

- CO₂ activation mechanism in eCO₂RR: SEPT mechanism at the common working U → CPET mechanism at the highly negative U
- The inverted region in the SEPT mechanism originates from the rapid rising of Pauli repulsion in the physisorption PEC as U decrease
- Effective designs of electrocatalysts can suppress the adverse effect of Pauli repulsion on the kinetics of CO₂ activation in eCO₂RR

1. L. Liu, H. Xiao, Inverted Region in Electrochemical Reduction of CO₂ Induced by Potential-Dependent Pauli Repulsion. *J. Am. Chem. Soc.* **145**, 14267–14275 (2023).
2. M. T. M. Koper et al. Advances and challenges in understanding the electrocatalytic conversion of carbon dioxide to fuels. *Nat. Energy* **4**, 732–745 (2019).